





## California State University of Bakersfield, Department of Chemistry

# **VB** Rocket



### **Standards:**

<u>K.PS2-1.</u> Plan and conduct an investigation to compare the effects of different strengths or different directions of pushes and pulls on the motion of an object.

<u>K.PS2-2.</u> Analyze data to determine if a design solution works as intended to change the speed or direction of an object with a push or a pull.

## **Introduction:**

Baking soda and vinegar are two common household items. Did you know that when they are combined they will create a forceful reaction? Baking soda and vinegar react with each other because of an acid-base reaction. Baking soda is a bicarbonate ( $NaHCO_3$ ) and vinegar is an acetic acid ( $CH_3COOH$ ). One of the products this reaction creates is carbon dioxide.

#### **Materials:**

- Film canister
- Rubber stopper/cork
- Baking soda
- Vinegar
- Scotch tape or masking tape

- Empty 2-liter soda bottle
- Construction paper (various colors)
- Scissors
- Paper towels

## Safety:

- Always have an adult with you to help you during your experiment.
- Always wear eye protection and gloves when doing chemistry experiments.
- Always perform this experiment outdoors in an open area.
- Always be cautious reaction can happen quickly.

#### **Procedure:**

- 1. Build rockets using film canisters and construction paper (we added wings and a nose to our canister to create a rocket).
- 2. Pour about 2 cups of vinegar into the empty 2-liter bottle.
- 3. Pour about ¼ cup of baking soda on to a paper towel and roll it tightly so no baking soda falls out. We used ½ of a full size paper towel. (The paper towel w/ baking soda needs to fit through the top of the 2-liter bottle)
- 4. Quickly put the paper towel w/ baking soda through the top of the 2-liter bottle.
- 5. **Firmly** place the cork/rubber stopper inside the top of the 2-liter bottle.
- 6. Place the rocket on top of stopper/cork.
- 7. Step back and watch the reaction. (It may take a few moments for it to react)

### **Data and Observations:**

1. Record your observations in this space.

## **Questions:**

2. What makes the rocket soot off?

### **References:**

1. <a href="http://www.stevespanglerscience.com/lab/experiments/acid-base-rocket">http://www.stevespanglerscience.com/lab/experiments/acid-base-rocket</a> (Date Accessed; August 4, 2014).